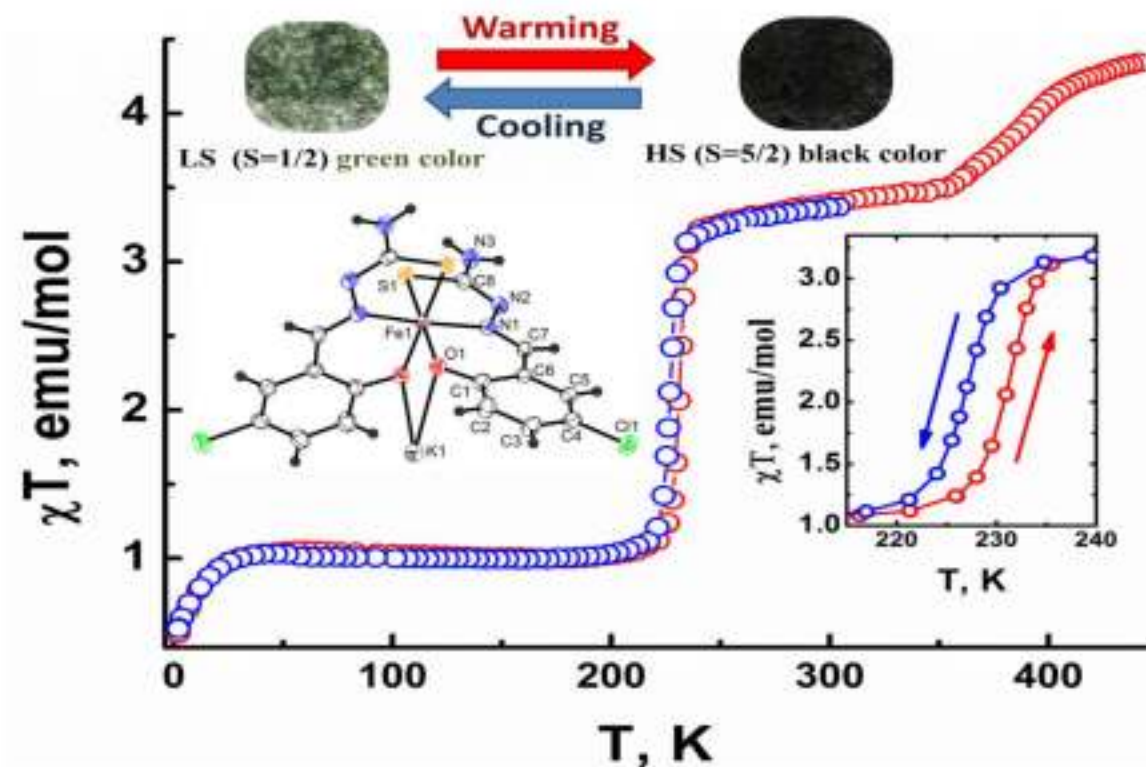


## Anionic spin-crossover complex of Fe(III) with space symmetry transition and thermal hysteresis around room temperature

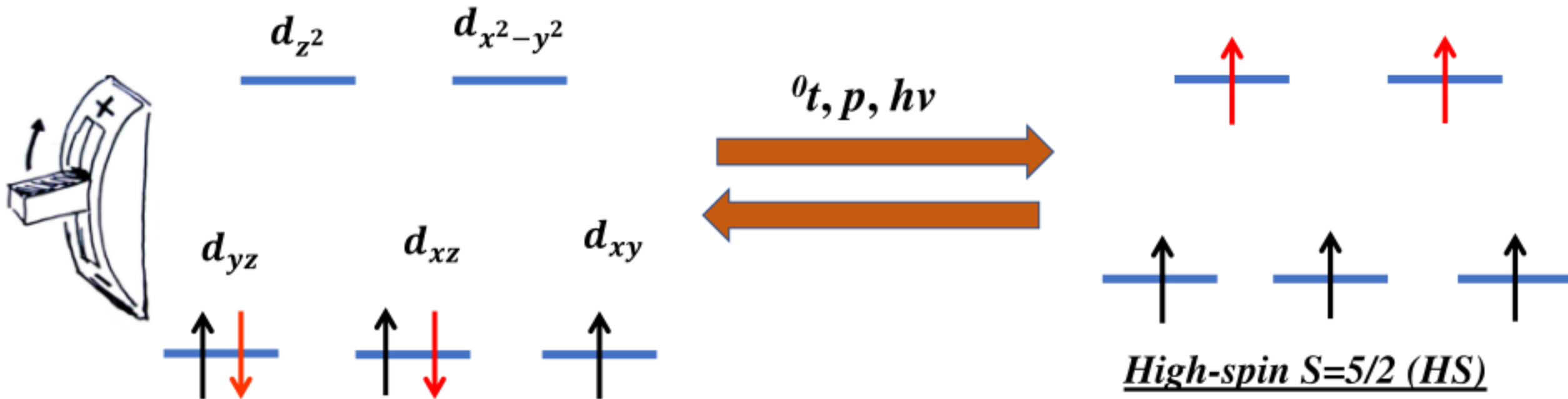
**Nataliya G. Spitsyna**, Maxim A. Blagov

*Institute of Problems of Chemical Physics, RAS, Semenov av.  
 1, 142432, Chernogolovka, Russia*

\* E-mail: [spitsyna@icp.ac.ru](mailto:spitsyna@icp.ac.ru)



# Spin Crossover Fe(III) with the $d^5$ electronic configuration



Low-spin  $S=1/2$  (LS)

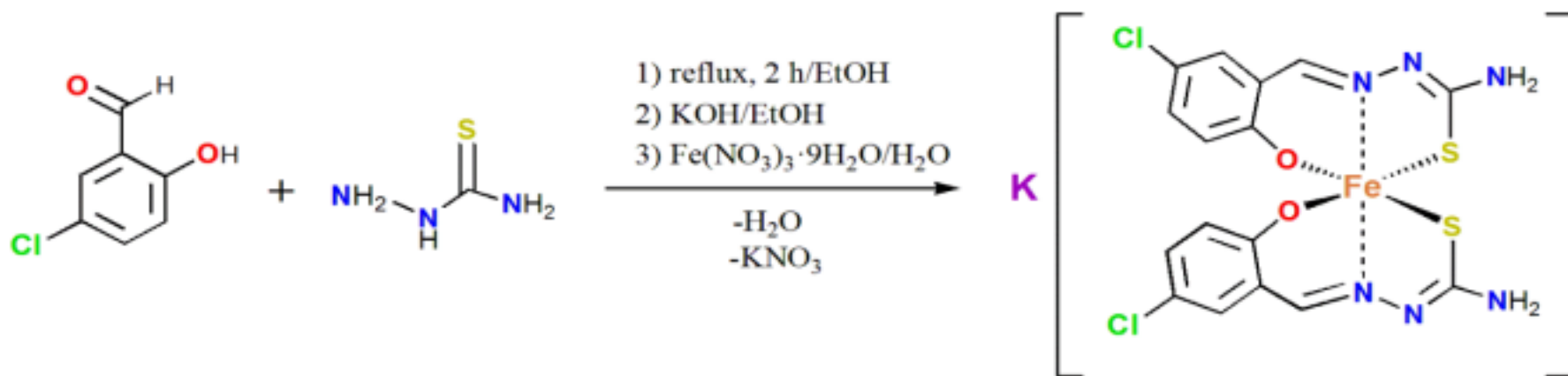
- Reversible SCO:  $S=1/2 \leftrightarrow S=5/2$ ;
- Changes in crystal structure: Fe-ligand bonds get elongation by  $\sim 10-15\%$  (relative to the LS state).

Potential applications:

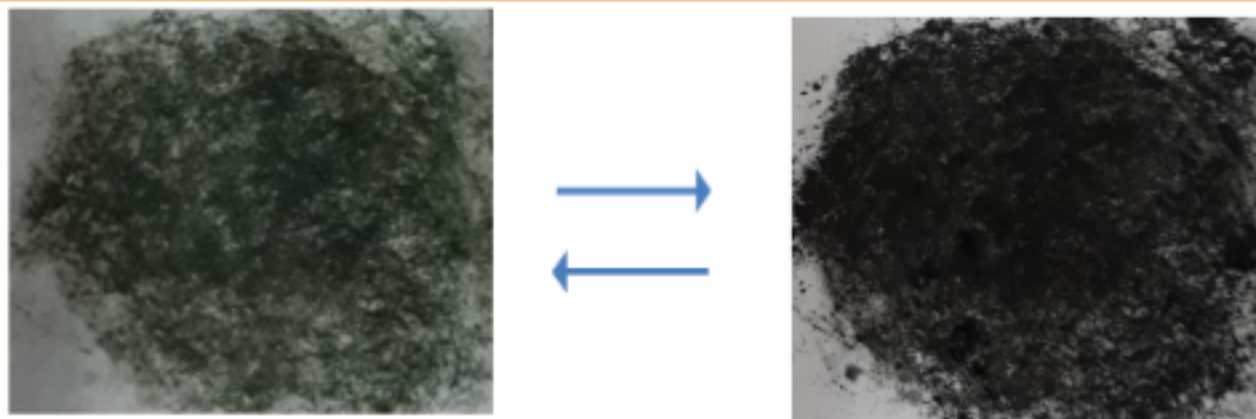
- Molecular switches;
- Data storage media;
- Nanoscale electronic devices;
- Biomimetic soft actuators.



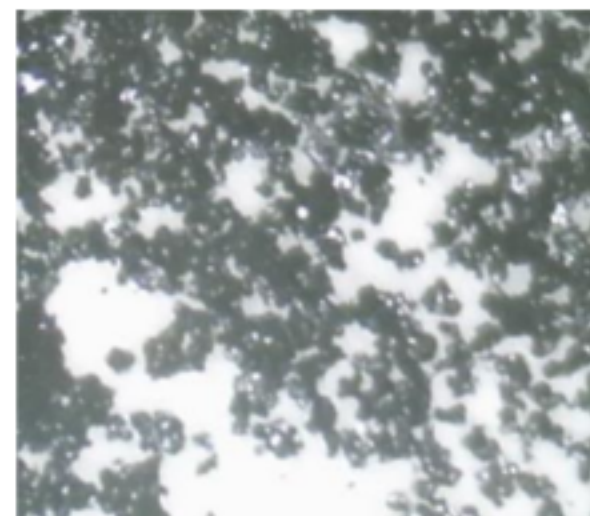
# Scheme of synthesis of the complex $K[Fe(5Cl-thsa)_2]$



A polycrystalline powder of  $K[Fe(5-Cl-thsa)_2]$  (**1**) was isolated.



**Photo of the thermochromism** for the polycrystalline sample of **1** was green color (*LS*,  $S=1/2$ ) at 77K, and black color (*HS*,  $S=5/2$ ) at 300K.

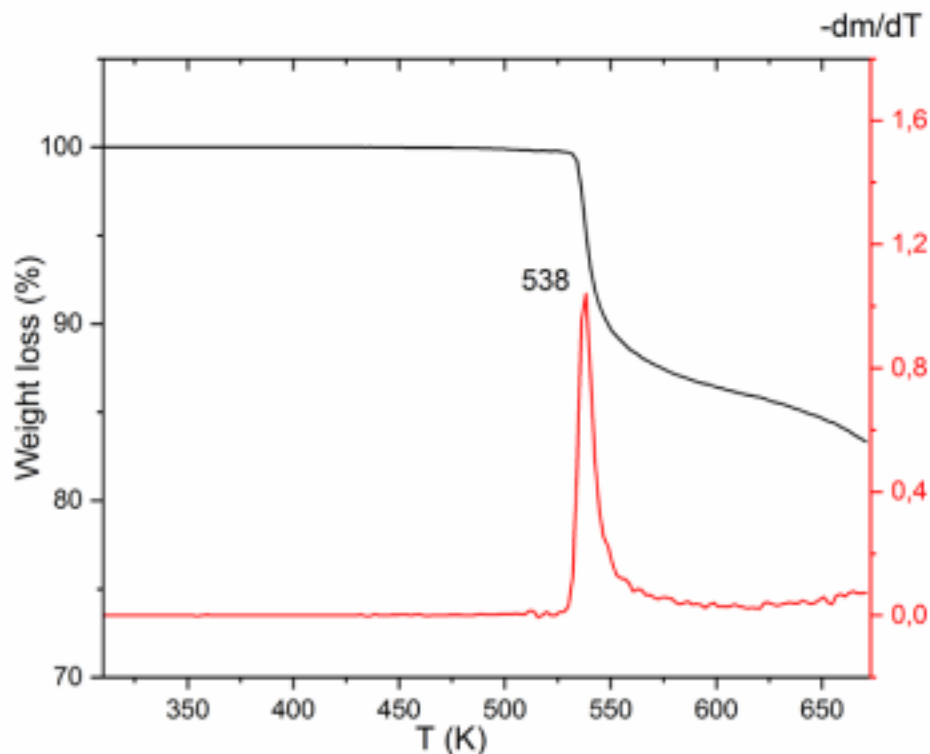


**Photo of the polycrystalline sample of  $K[Fe(5Cl-thsa)_2]$ .**

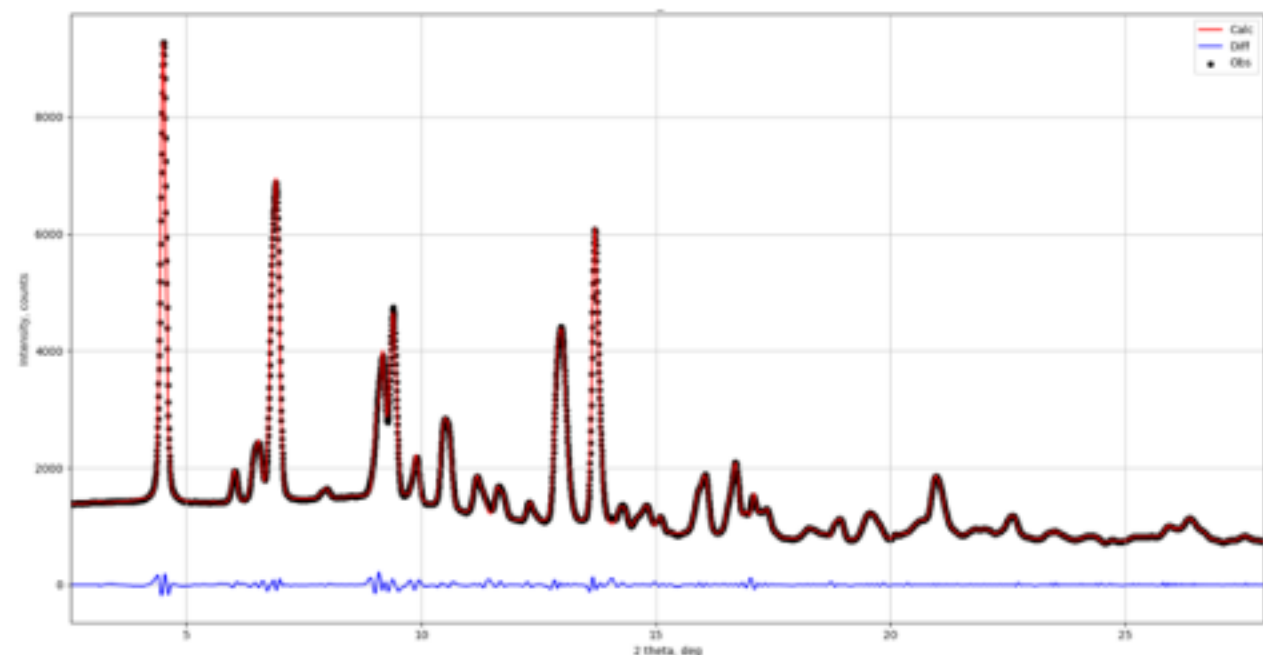




# Characterization of $\text{K}[\text{Fe}(\text{5Cl-thsa})_2]$ salt



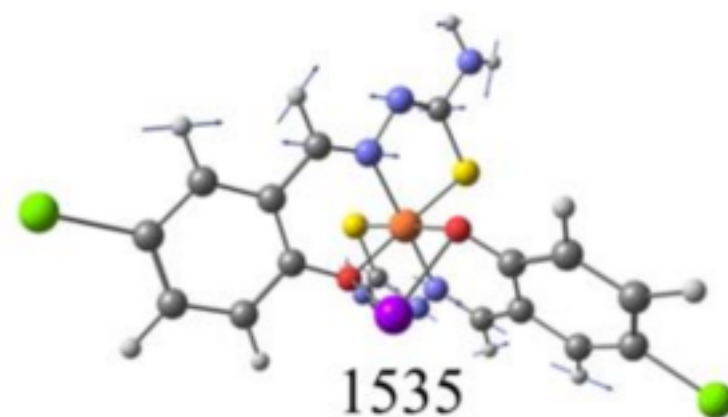
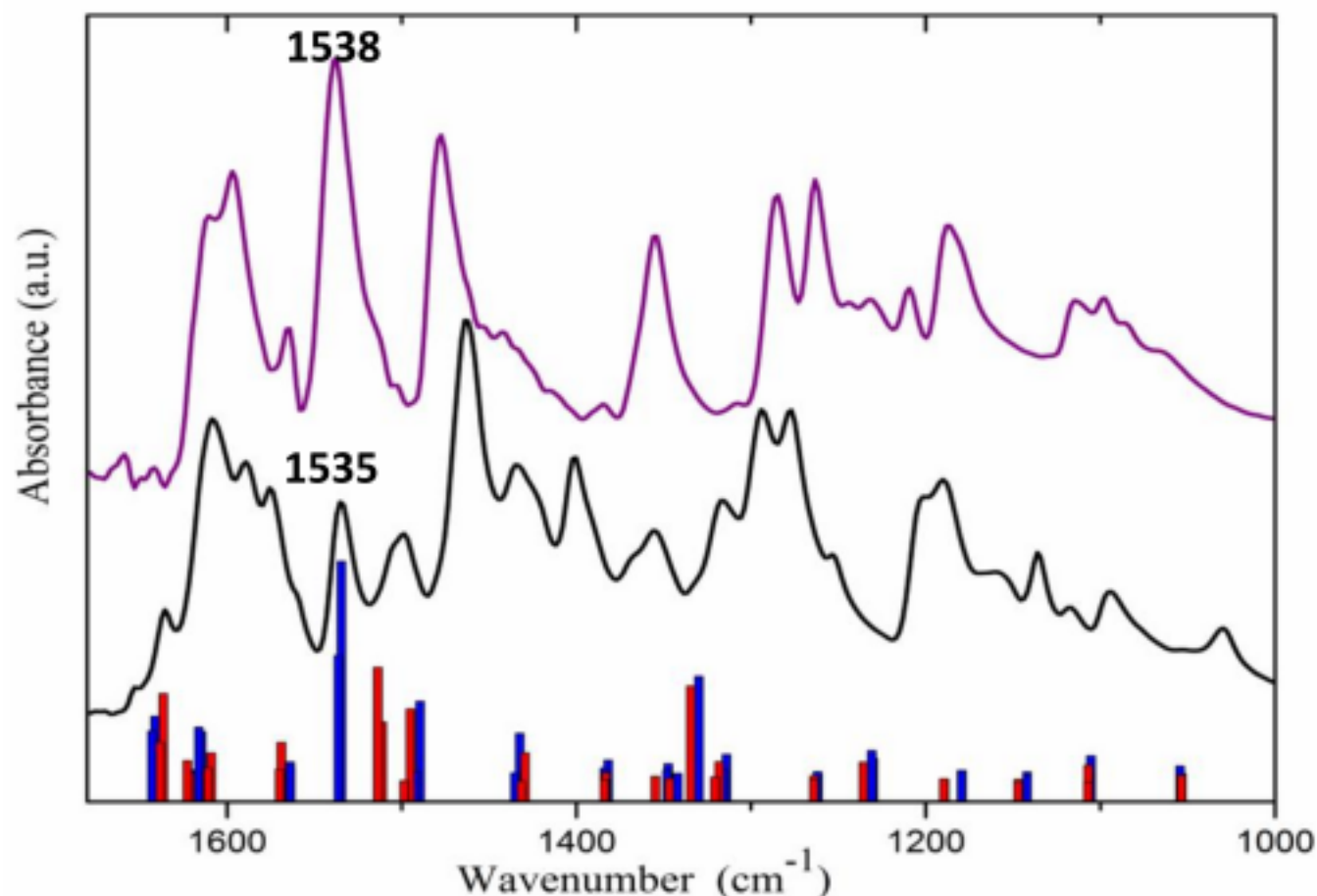
**The thermogravimetric analysis** for  $\text{K}[\text{Fe}(\text{5Cl-thsa})_2]$  was performed using a NETZSCH STA 409 C Luxx thermal analyzer. Sample weight - 8.366 mg, atm.- argon, flow rate  $V=12$  ml/min, temperature range- 27 - 700 °C, heating rate 10 °C/min.



**Powder X-ray diffraction pattern** of  $\text{K}[\text{Fe}(\text{5Cl-thsa})_2]$  taken at room temperature (black dots) and LeBail method refinement of XRPD data (red line) for orthorhombic  $P_{bcn}$  structure. The bottom line is the difference plot.

**Elemental analysis.** Anal. Calcd (%) for  $\text{K}[\text{Fe}(\text{5Cl-thsa})_2]$ ,  $\text{C}_{16}\text{H}_{12}\text{Cl}_2\text{FeKN}_6\text{O}_2\text{S}_2$  (550.28  $\text{g}\cdot\text{mol}^{-1}$ ): C, 34.92; H, 2.20; N, 15.27; S, 11.63. Found: C, 34.99; H, 2.57; N, 15.24; S, 11.28 %. The electron-probe X-ray microanalysis afforded the elements ratio is Fe:K:S=1:1:2. The complex is soluble in water, ethanol, acetonitrile and DMF.

# IR Spectroscopic studies

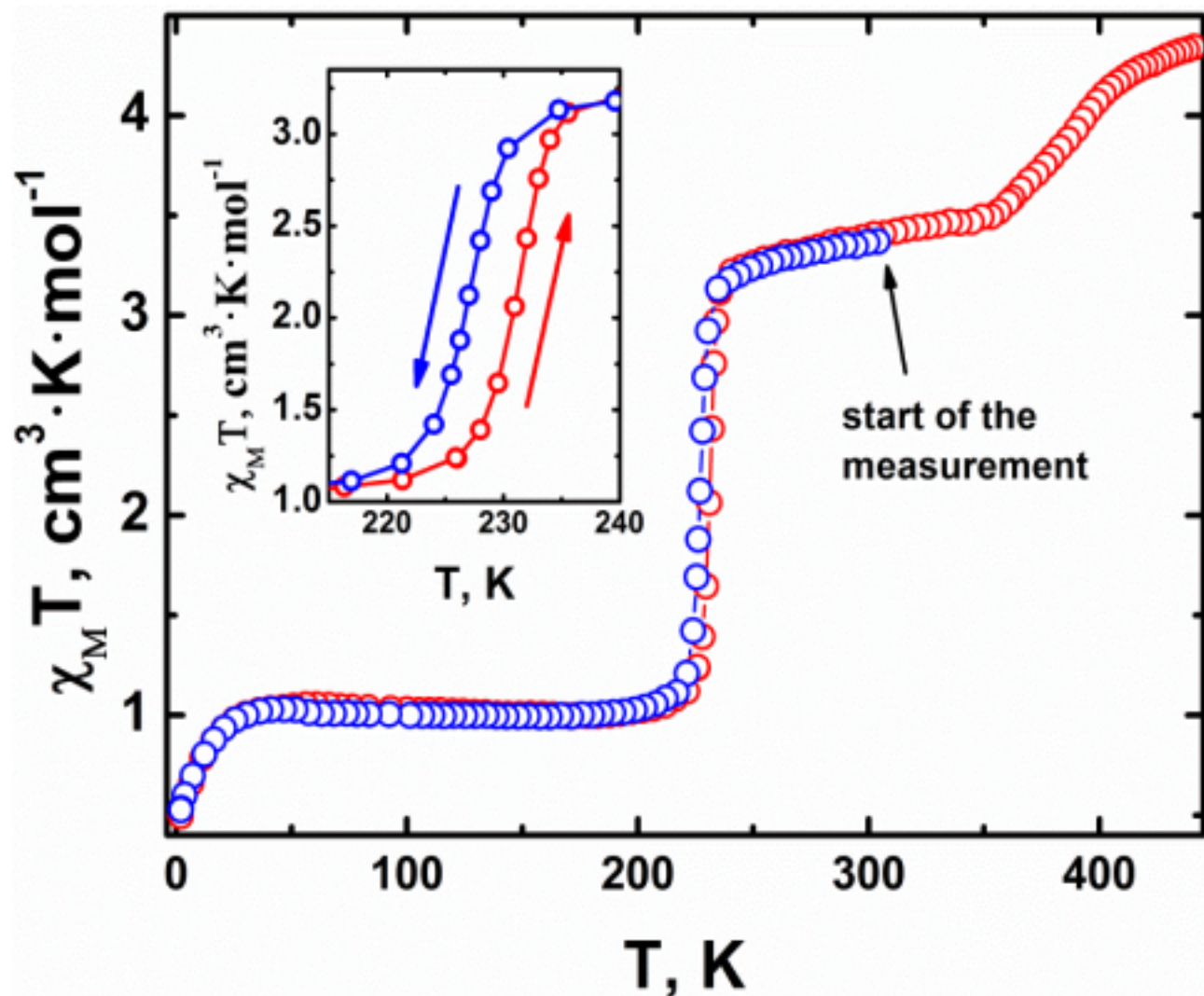


- According to the calculated geometry of the cation and the IR spectrum, salt **1** is in the LS state
- The intense band (arrow) at 1535 cm<sup>-1</sup> (C=Nim– stretching vibrations) shows the existence of a magnetically active ion configuration at 298K *LS*, while according to the magnetic data,  $\gamma_{HS}$  is **78%**.

**Experimental ATR FT-IR** absorption spectra for the H<sub>2</sub>-5Clthsa (violet line) and the K[Fe(5Cl-thsa)<sub>2</sub>] (black line) at T=298K, in the range 1600–1000 cm<sup>-1</sup>. IR vibration frequencies calculated by DFT are given as bars at the bottom part of the Figure for K[Fe(5Cl-thsa)<sub>2</sub>] in the HS (red) and LS (blue) states. The B3LYP\* functional with the 6-31G(d,p) basis set was used.



# Bulk magnetic properties



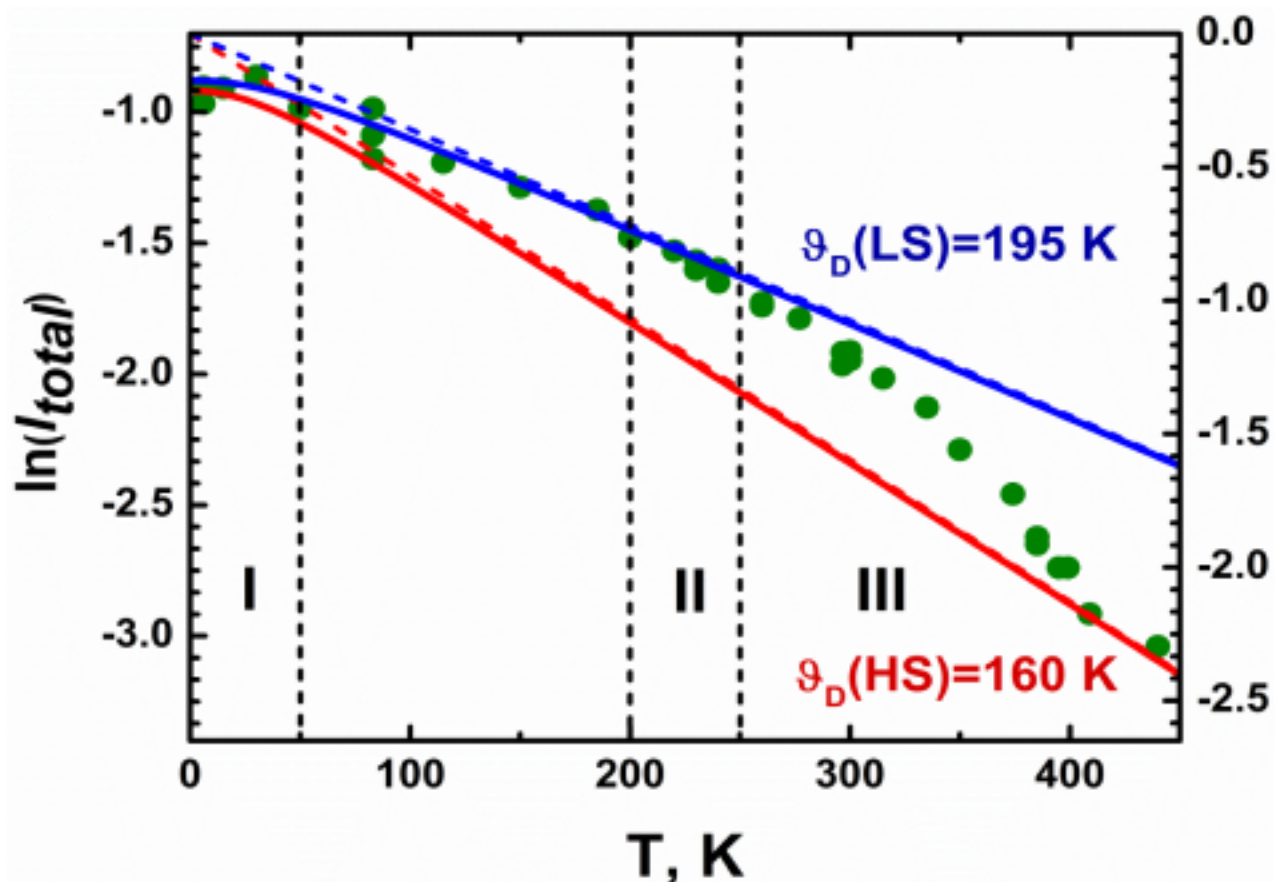
Temperature dependences of  $\chi_M T$  product in the heating (red circles) and cooling (blue circles) modes. The scan rate is 2 K/min.

- The magnetic susceptibility for the **K[Fe(5Cl-thsa)<sub>2</sub>]** salt has a complex temperature dependence. It can be interpreted as a three-step SCO from 96% of LS at 2K to 100% of HS fraction at 440K.
- The effective half-transition temperatures for **1** are  $T_{1/2\downarrow}=228$  K and  $T_{1/2\uparrow}=232$  K ( $T_{1/2}=\frac{1}{2}\cdot(T_{1/2\uparrow}+T_{1/2\downarrow})=230$ K);
- A hysteresis of magnetic response with a maximum width of  $\sim 6$  K is observed in a temperature range of 215-235K.
- ❑ First step – the conversion of residual HS fraction from 4% to 17%. (In the temperature range of 2- 50K)
- ❑ Second step – an abrupt spin transition increase in  $\gamma_{HS}$  from 17% to 74%. (In the temperature range of 200 - 250K,  $\Delta\gamma_{HS}=57\%$ )
- ❑ Third step– increase in  $\gamma_{HS}$  to essentially 100% ( $\Delta\gamma_{HS}=26\%$ ) between 250 - 440K





# $^{57}\text{Fe}$ Mössbauer spectroscopy of $\text{K}[\text{Fe}(\text{5Cl-thsa})_2]$



**Temperature dependences** of the total spectrum intensity  $\ln(I_{total})$  logarithm and the logarithm of the Mössbauer effect probability  $\ln(f)$ .

Solid lines - dependences for specific values of the Debye temperature

$\ln(I_{total}(T)) = -\frac{3E_R^2}{k_B\theta_D^3} \int_0^{\theta_D} x \cdot \text{cth}\left(\frac{x}{2T}\right) dx + \text{cons}$ , dashed lines - the classical approximation ( $T \gg \theta_D$ ).

## Determination of the Debye temperatures for the *HS* and *LS* states of a magnetically active anion $\text{Fe}(\text{III})$

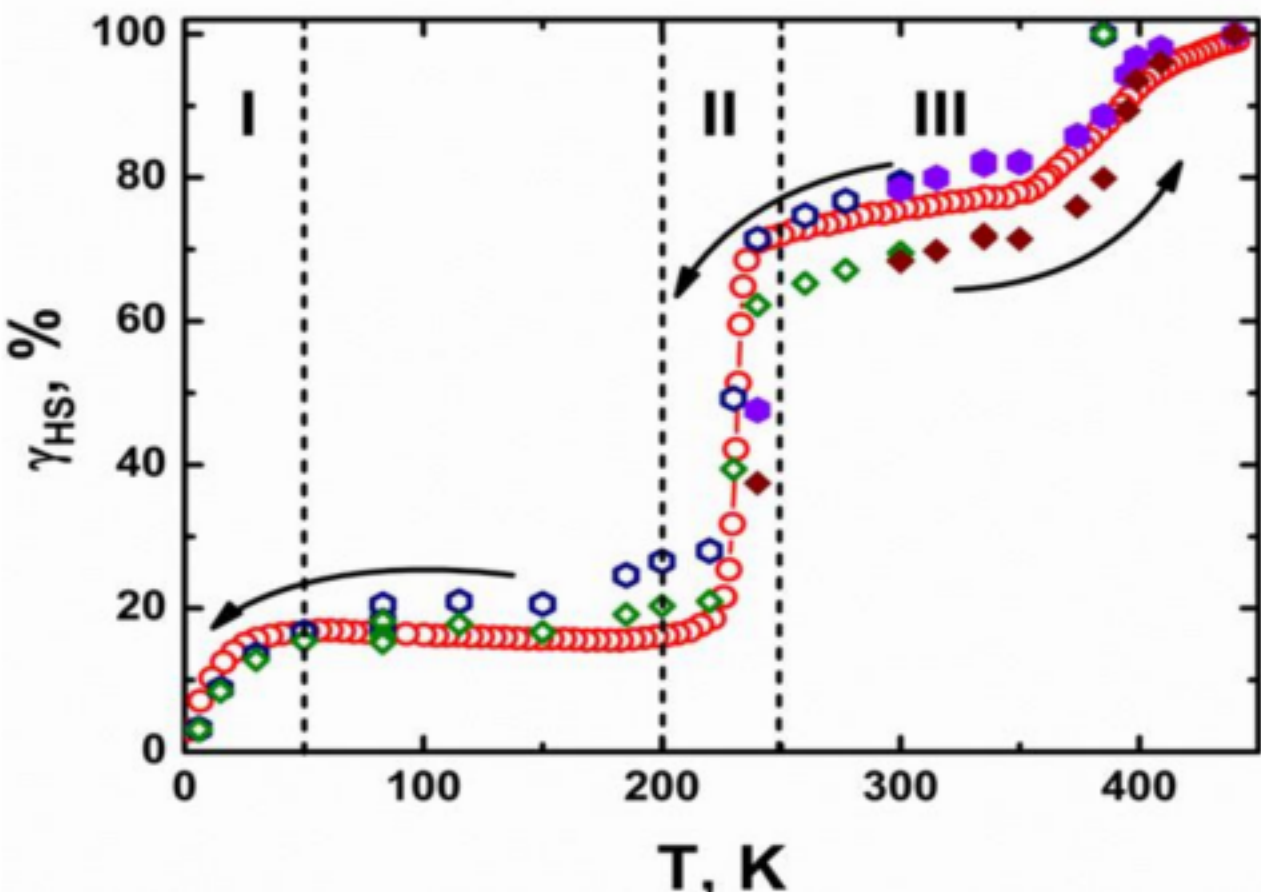
- Debye temperatures ( $\theta_D$ ) for *LS* and *HS* complex were obtained from the Mössbauer data with the aim of the precise calculations of  $\gamma_{HS}$  values at different temperatures which corresponds to all steps of SCO from only *LS* state up to only *HS* state:

$$\gamma_{HS} = \frac{I_{HS}(T) \cdot f_{LS}(T)}{I_{HS}(T) \cdot f_{LS}(T) + (100 - I_{HS}(T)) \cdot f_{HS}(T)} \cdot 100\%$$

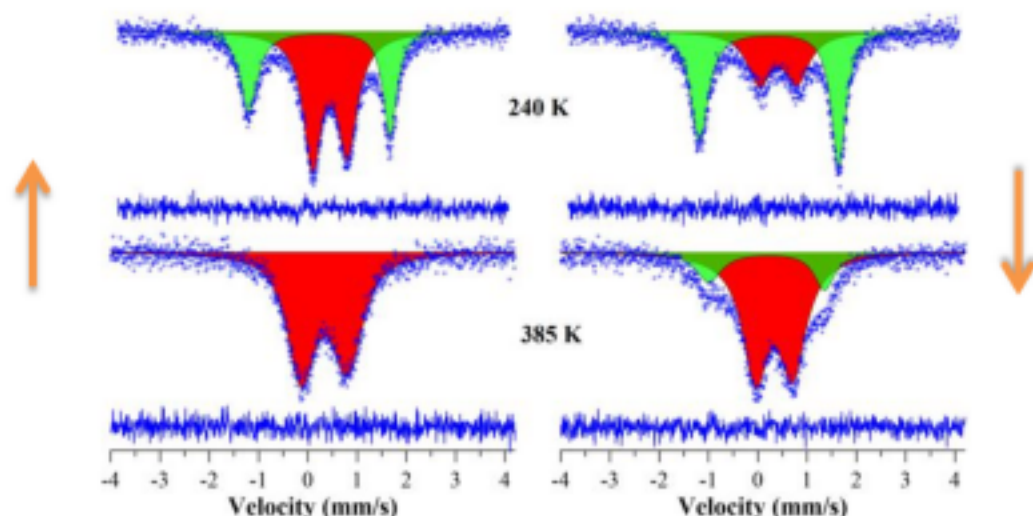
- The transition from the *LS* Debye temperature to the *HS* Debye temperature begins only at stage **III** at 250K, when the proportion of the *HS* fraction reaches 78%, and ends at 400K, when the content of the *HS* fraction is 90% - the "delay" effect.



# $^{57}\text{Fe}$ Mössbauer spectroscopy



**Temperature dependence** of *HS* state fraction as determined from Mössbauer experiment before (rhombuses)/after (hexagons) correction with Debye temperature and *DC* magnetic measurements (red circles) data of  $\text{K}[\text{Fe}(\text{5Cl-thsa})_2]$  major polymorph. Empty blue hexagons/green rhombuses correspond to cooling mode, and violet hexagons/brown rhombuses correspond to heating mode.



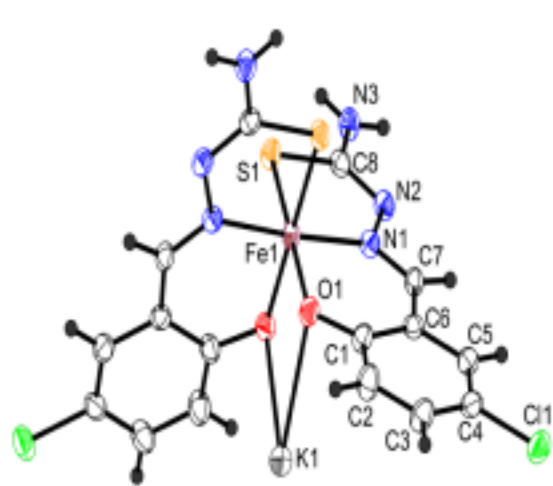
**Mössbauer spectra of major polymorph  $\text{K}[\text{Fe}(\text{5Cl-thsa})_2]$  taken at various temperatures.**

- Mössbauer data almost perfectly match with SQUID data.
- Mössbauer data after correction for  $\vartheta\text{D}$  become closer to magnetic data at high temperature range (225-440K) but at low temperature range (50-225 K) match worse.
- *HS* fraction at 240K is either  $\sim 49\%$  or  $\sim 71\%$  as obtained in the heating and cooling modes, respectively. At 385K, the *HS* fractions are 88% and  $\sim 100\%$  upon heating and cooling respectively.

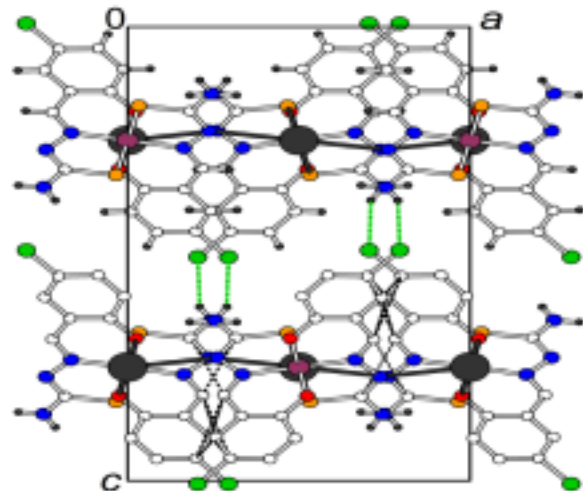


# X-ray Crystallography of $\text{K}[\text{Fe}(\text{5Cl-thsa})_2]$

Minor polymorph with ordered potassium cations



The molecular and crystal structure of 1.



Major polymorph with disordered potassium cations

?

The experimental powder pattern was successfully indexed assuming an orthorhombic lattice (with systematic absences conforming with the  $Pbcn$  space group) to yield the following parameters (the values in brackets correspond to analogous parameters from the single-crystal experiment at the same temperature of 100 K):

$$a = 11,4131(11,615) \text{ \AA};$$

$$b = 8,7513 (8,3480) \text{ \AA};$$

$$c = 20,2123 (20,763) \text{ \AA};$$

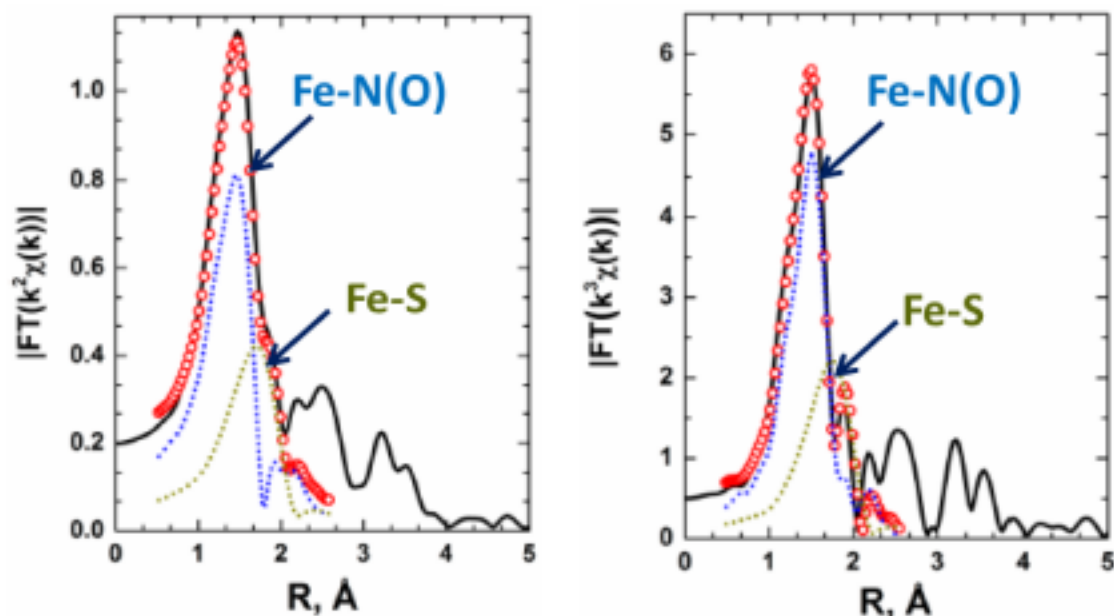
$$V = 2018,8 (2013,2) \text{ \AA}^3.$$

300 K

Single-crystal synchrotron X-ray diffraction studies of  $\text{K}[\text{Fe}(\text{5Cl-thsa})_2]$  at 100K, 250K and 300K revealed the same orthorhombic  $Pbcn$  space group. According to  $PXRD$ , it is absent in the main phase. Dimensions of the found single crystal (mm):  $0.05 \times 0.03 \times 0.02$ .



# Synchrotron-based *EXAFS* investigation



**Fourier transforms of the Fe K-edge *EXAFS* spectrum of  $K[Fe(5Cl-thsa)_2]$** , corresponding to two different weighting schemes  $k^2\chi(k)$  (left) and  $k^3\chi(k)$  (right) giving rise to varied relative intensities of the Fe-N(O) and Fe-S components: experimental data (black lines), total theoretical best-fit (red circles), Fe-N(O) component (blue circles), Fe-S component (swamp-green circles).

**Best-fit local-structure parameters around Fe atoms in  $K[Fe(5Cl-thsa)_2]$  according to *EXAFS* at 300 K.**

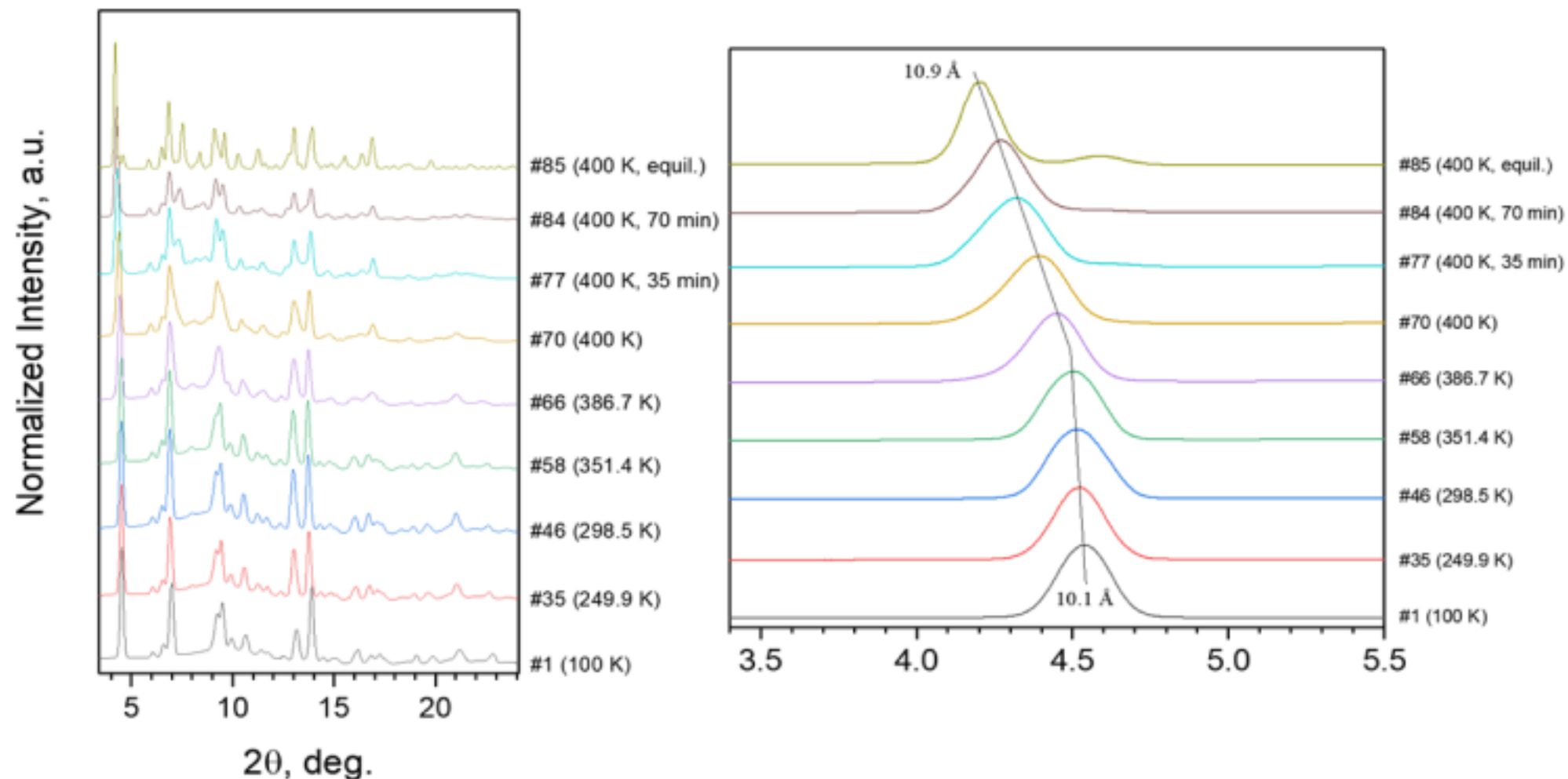
Path	Coordination number N	Interatomic dist. R, Å	Debye-Waller factor $\sigma^2$ , Å <sup>2</sup>	$\Delta E$ , eV	$R_f$
Fe-N(O)	4	1.95	0.0079	-6.6	0.010
Fe-S	2	2.28	0.0169		

Fitting ranges:  $k=2.5 \dots 15.0 \text{ \AA}^{-1}$ ,  $R=1.1 \dots 2.2 \text{ \AA}$ .

- The bond lengths of the coordination octahedron for both polymorphs are identical and corresponds LS structure:
  - *X-ray*  $l(Fe-O)=1,953(2)\text{ \AA}$ ,  $l(Fe-N)=1,951(2)\text{ \AA}$ ,  $l(Fe-S)=2,2390(7)\text{ \AA}$ ;
  - *EXAFS*  $l(Fe-O/N)=1,95 \text{ \AA}$ ,  $l(Fe-S)=2,28 \text{ \AA}$  ( $\Delta \leq 0,041 \text{ \AA}$ ).
- There is an effect of structural «laggin» for the main polymorph (powder), since at room temperature the proportion of the HS fraction is 75%.



# Synchrotron-based XRPD investigation

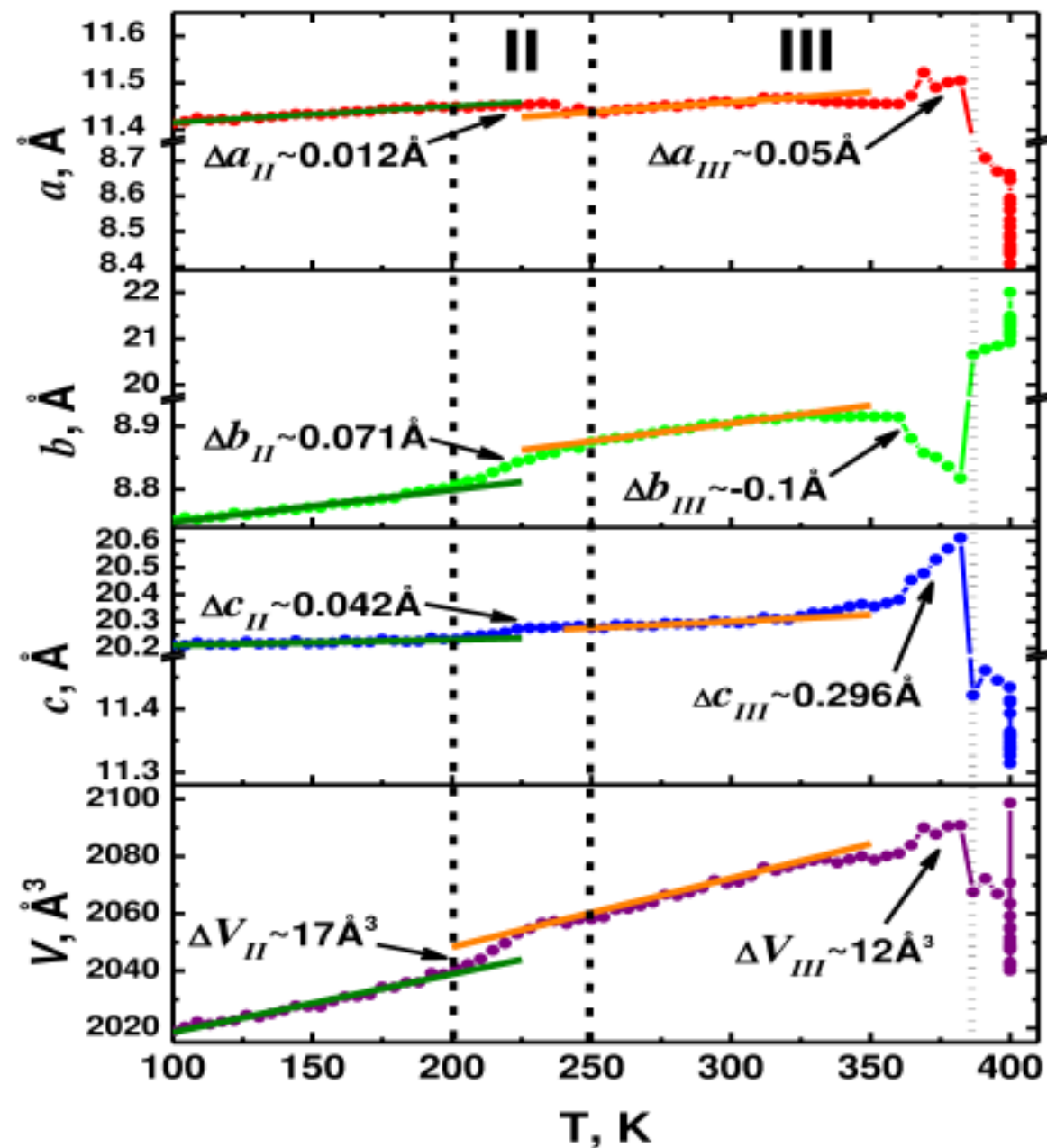


## Powder diffraction patterns at various temperatures for $\text{K}[\text{Fe}(\text{5Cl-thsa})_2]$ ( $\lambda=0.8000 \text{ \AA}$ ).

The right panel shows the change in temperature of the first low-angle peak corresponding to the interplanar spacing of the layered structure. The general structure of peaks experiences only minute changes on going from 100K (pattern #1) up to 386.7 K (pattern #66).



# Synchrotron-based XRPD investigation



The changes of unit cell parameters  $a$ ,  $b$ ,  $c$  and  $V$  (based on XRPD) data in the temperature range 100-400 K for  $\text{K}[\text{Fe}(\text{5Cl-thsa})_2]$  salt.

Roman numerals II ( $\gamma_{\text{HS}}$  from 17% to 75%,  $\Delta \gamma_{\text{HS}}=58\%$ ) and III ( $\gamma_{\text{HS}}$  75 to 100%,  $\Delta \gamma_{\text{HS}}=25\%$ ) indicate the SCO step numbers according to magnetic measurements.

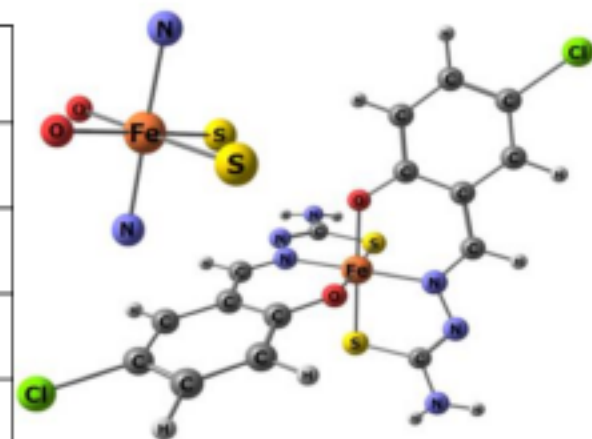
The short dashed lines (---) mark borders of the distinct SCO steps. The dotted line ( $\cdots$ ) at  $\sim 387 \text{ K}$  indicates the orthorhombic-to-monoclinic structural phase transition with a symmetry lowering from orthorhombic to monoclinic ( $Pbcn \rightarrow P21/n$ ) space group.

Note that the Y-axes for  $a$ ,  $b$ , and  $c$  parameters contain breaks.

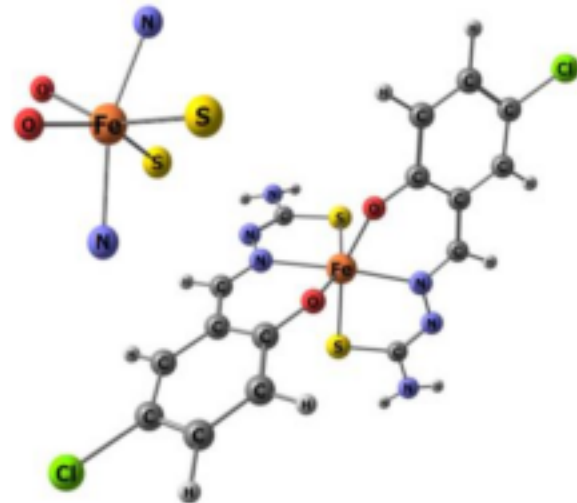


# Theoretical calculations

no.	Functional	Type	HF, %	Spin state	$E_{el}$ , a.u.	$E_{ZPV}$ , a.u.	$E_0$ , a.u.	$\Delta E_0(\text{HS-LS})$ , kJ/mol
1	OLYP	GGA	—	HS	-4676.896743	0.267317	-4676.629426	6.34
				LS	-4676.901767	0.269926	-4676.631841	
2	OPBE	mGGA	—	HS	-4676.519302	0.269249	-4676.250053	25.78
				LS	-4676.531937	0.272064	-4676.259873	
3	TPSS	GH-mGGA	10	HS	-4677.2150951	0.267466	-4676.9476291	85.67
				LS	-4677.2503820	0.270123	-4676.9802590	
4	TPSSh	GH-mGGA	10	HS	-4677.020263	0.271916	-4676.748347	44.40
				LS	-4677.039775	0.274515	-4676.76526	
5	PBE0-15	GH-GGA	15	HS	-4674.5620255	0.271981	-4674.2900445	18.97
				LS	-4674.571944	0.274673	-4674.2972710	
6	B3LYP*	GH-GGA	15	HS	-4675.6142460	0.270846	-4675.3434000	20.07
				LS	-4675.6244784	0.273434	-4675.3510444	
7	B3LYP	GH-GGA	20	HS	-4676.9703525	0.272984	-4676.6973685	-2.04
				LS	-4676.9720982	0.275505	-4676.6965932	
8	PBE0	GH-GGA	25	HS	-4674.6515481	0.276317	-4674.3752311	-20.36
				LS	-4674.6464029	0.278928	-4674.3674749	



***LS, S=1/2***



***HS, S=5/2***

The OLYP, OPBE, B3LYP\* and PBE0-15 functionals tested in DFT calculations confirm the feasibility of the SCO transition. A parametrization of the B3LYP functional was carried out in order to determine the optimal proportion of the Hartree-Fock exchange.

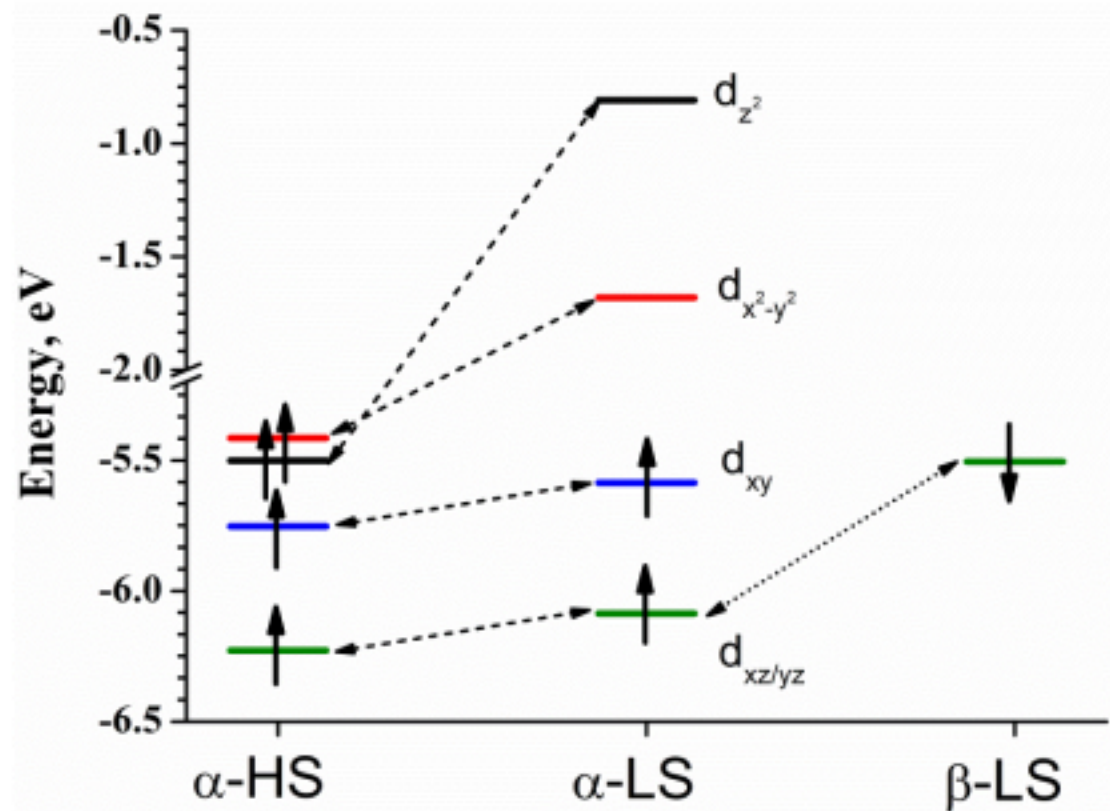
Optimization and calculation of frequencies: 6-31G(d,p);

Calculation of electronic energy  $E_{el}$ : 6-311+G(2df,2p);

Quantum Chemistry Package: Gaussian 09.

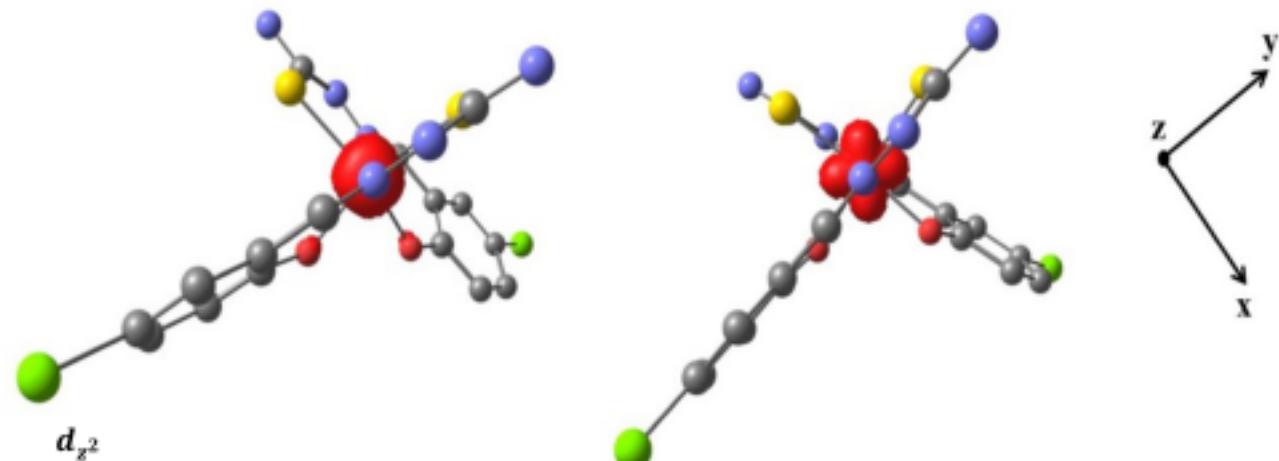


# Theoretical calculations



## The energy level diagram

of molecular orbitals with a predominant iron  $d$  orbitals contribution for the  $HS$  and  $LS$  states of the  $K[Fe(5Cl-thsa)_2]$  complex. Note that the Y-axis contains break. (B3LYP\*/6-31G(d,p)).



**Estimated spin density distribution**  
for  $HS$  (left) and  $LS$  (right) structures of  $K[Fe(5Cl-thsa)_2]$ .

DFT calculations highlight the special role played by  $\sigma$ -bonding between the  $N_{im}$  ligand sites and the  $d_{z^2}$  orbital of the iron atom, which is consistent with the maximum relative elongation of the Fe- $N_{im}$  bonds for all thsa-based complexes.





## Concluding remarks :

- ✓ 2D heterometallic polymer  $\mathbf{K[Fe(5Cl-thsa)_2]}$  (**1**) has been synthesized and thoroughly characterized. The existence of two polymorphs for **1**, with ordered and disordered potassium cations, was found. According to the *PXRD*, only “trace” amounts of a polymorph with an ordered **K** cation are present in the main phase. The formation of a poorly crystallizing basic polymorph explains why it is difficult to obtain single crystals for thsa- complexes. The established structure of  $\mathbf{K[Fe(5Cl-thsa)_2]}$  on a single crystal is the 4-th one in CCDC since the 1970s, when these complexes were discovered.
- ✓ DC magnetization, Mössbauer measurements consistently indicate that the second major polymorph with potassium cations disordered over a few crystallographic sites undergoes a complicated three-step cooperative spin-crossover transition in a temperature range 2-440 K involving the following steps: I (2-50 K, gradual increase in the HS fraction), II (200-250 K, an abrupt increase in the HS fraction), III (250-440 K, gradual increase in the HS fraction to 100%). The SCO curve has a half transition temperature  $T_{1/2}=230$  K and a hysteresis loop with a width of 6 K.



## Concluding remarks:

- ✓ According to EXAFS, Mössbauer- and IR- spectroscopy data SCO for the dominant polymorph is quite peculiar. The effect of "structural delay" of the rearrangement of the LS into the HS state is shown. Indeed, the increase in the HS concentration by 57% at the second step does not result in the expected significant increase in the iron(III)-ligand bond lengths. In addition, the final step of the spin conversion ( $\Delta\gamma_{HS}=26\%$ ) is associated with a structural phase transition with a symmetry lowering from orthorhombic (*Pbcn*) to monoclinic (*P21/n*) space group.
- ✓ These results provide a new platform for understanding the multistep SCO character in the Fe(III) thsa-complexes and related compounds. The features of the electronic structure are established.



## We thank:

Yan V. Zubavichus, Vladimir A. Lazarenko,  
Roman D. Svetogorov – the **SRF SKIF Boreskov IC  
SB RAS and NRC “Kurchatov institute”** for

carrying out structural measurement at synchrotron *X-ray* structural analysis beamline.



Alexander N. Vasiliev, Olga Maximova, Vyacheslav S. Rusakov, Sergey A. Yaroslavtsev –  
**Lomonosov Moscow State University Faculty of Physics** for the *dc* – magnetometry  
measurements and Mossbauer spectral investigations.

**ICCP RAS** – Vladimir B. Krapivin, Maxim A. Blagov performed DFT calculations; – Vera P. Lodygina, Anatolii S. Lobach measured and contributed to analyses IR and TGA data.

This work was partially done with using of the  
Computational and Analytical Center  
for Collective Use of the **ICCP RAS** tool base.

(State task, state registration No. AAAA-A19-119092390079-8)





THANK YOU FOR  
ATTENTION!!

