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**Effect of prescription configuration  
on properties nanothermite composition  
 $\text{Bi}_2\text{O}_3/\text{Al}/1\text{Me-3H}$**

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# NANOTHERMITE



e.g.:  $\text{Fe}_2\text{O}_3/\text{Al}$ ,  $\text{CuO}/\text{Al}$ ,  $\text{MoO}_3/\text{Al}$  etc.

## Peculiar properties:

- High sensitives to mechanical and electrical influences, which are at the sensitivity level initiating explosive;
- Low critical parameters of combustion performance – burn in thin layer (0.1 mm and below), initiation in microgram quantities;
- The burn rates nanothermites can vary from cm/s to hundreds of m/s, regulated by prescription and technological parameters.

## Application:

- initiation systems;
- fast-burning compound;
- MEMS pyrotechnics.

# MATERIALS AND PREPARATION OF NANOTHERMITE COMPOSITES

**Bi<sub>2</sub>O<sub>3</sub>** (Sigma Aldrich, USA), a mean particle size of 90-210 nm, main component contain 99.8 %

**Al** (Advanced Powder Technologies LLC, Russia), a mean particle size of 90-150 nm with 85 % active aluminum

1-Methyl-3-nitro-1,2,4-triazole (**1Me-3H**) insensitive high-energy compound with empirical formula C<sub>3</sub>H<sub>4</sub>N<sub>4</sub>O<sub>2</sub>, enthalpy of formation  $\Delta H = 1.42$  kJ/g, melting temperature  $T_m \approx 65$  °C, and decomposition temperature  $T_d \approx 240-250$  °C.

## Preparation nanothermite composites:

- Pre-insertion processing of the components: preliminary deagglomeration of nanopowders, preparation of suspension of the additive in organic solvent (acetone)
- Ultrasonic intermixing of the powders in the suspension of the additive
- Suspension drying
- Grinding of the pyrotechnic mass beneath the layer of the volatile liquid that is not a solvent or plasticizer for a additive (hexane)
- Vacuum-assisted drying of the suspension to obtain a composite representing powdered agglomerated particles of the target composition

# METHODS

- The thermodynamic parameters of combustion of nanothermite composites (heat  $Q$ , combustion products pressure  $P$ ) were calculated using the REAL software package. Conditions of adiabatic combustion of a nanothermite charge were simulated in a confined space.
- Relative explosion force  $F$  (%) was determined by measuring the amplitude of the signal recorded by an oscillograph when the sample was initiated in a stain gauge sensor (fig. 1). For 100 % of magnitude  $F$  (basic level) corresponds to the explosion of the nanothermite pair  $\text{Bi}_2\text{O}_3/\text{Al}$ .
- The burning rate was determined by the ionization method using two configurations of charges: tube (fig. 2) and thin layer (fig. 3)

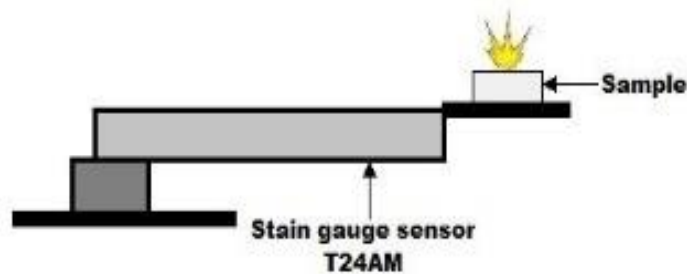


Fig. 1 – Test scheme for determining explosion force

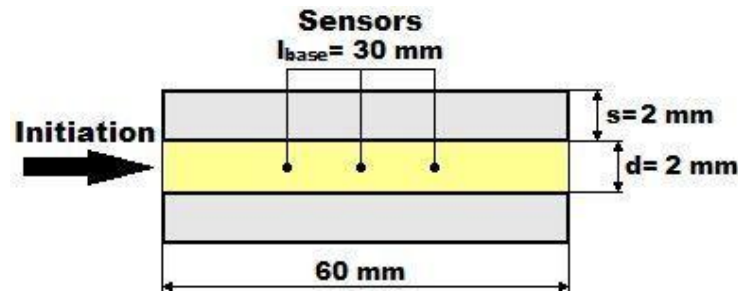


Fig. 2 – Test scheme for determining the burn rate in the tube

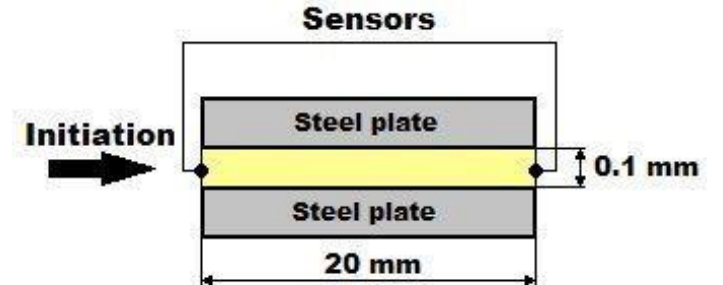


Fig. 3 – Test scheme for determining the burn rate in the thin layer

# **PRESCRIPTION CONFIGURATION**

**Configuration I** – the ratio of the components of the mixture corresponded to the maximum calculated heat of explosion of the composition ( $Q$ ) at a given content of 1Me-3H.

**Configuration II** – the ratio of the components of the mixture corresponded to the maximum calculated pressure value ( $P$ ) developed during combustion composition in a closed volume.

**Configuration III** – the ratio of the components of the base nanothermite pair  $\text{Bi}_2\text{O}_3/\text{Al}$  (88/12%), corresponding to the maximum calculated value of  $Q$  and remained constant in the mixture.

**Configuration IV** – the ratio of nanothermite components of the base nanothermite pair  $\text{Bi}_2\text{O}_3/\text{Al}$  (86/14%), corresponding to the maximum calculated value of  $P$  and remained constant in the mixture.

## RELATIVE EXPLOSION FORCE OF THE COMPOSITION $\text{Bi}_2\text{O}_3/\text{Al}/1\text{Me-3H}$

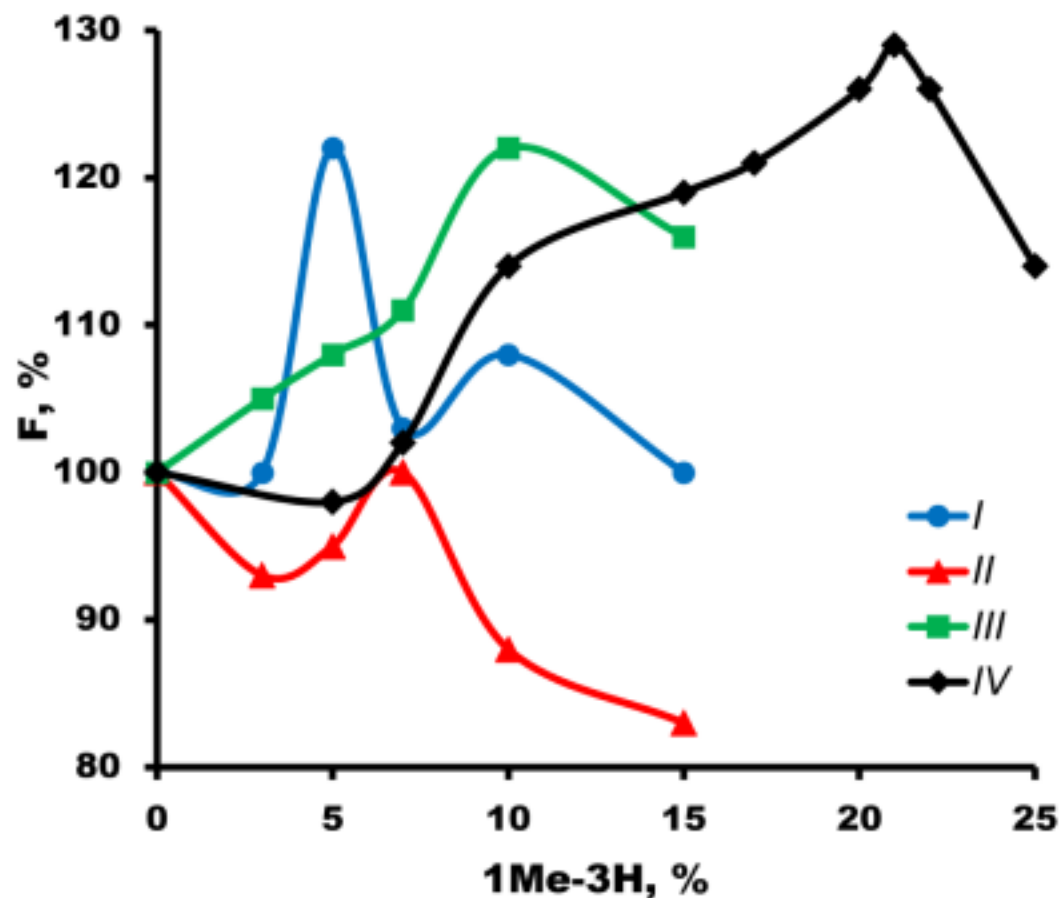


Fig. 4 - Relative explosion force  $F$  at different contents of additives

Configuration I, III, IV are characterized by the extreme nature of the dependence  $F$  on the content 1Me-3H.

Maximum  $F$  is observed at the following concentrations 1Me-3H:

Configuration I – 5 % 1Me-3H –  $F = 122$  %;

Configuration III – 10 % 1Me-3H –  $F = 122$  %;

Configuration IV – 21 % 1Me-3H –  $F = 129$  %.

In the case configuration II, with a content of 1Me-3H – 3, 5 %, there is a decrease in  $F$  to 93, 95 %. When the additive content is 7 %, the explosion force returns to the level of the base nanothermic mixture.

# BURNING RATES NANOTHERMITE COMPOSITION $\text{Bi}_2\text{O}_3/\text{Al}/1\text{Me-3H}$

	Component ratio, %	Density, % TMD	Burning rate, m/s	
			tube	layer
$\text{Bi}_2\text{O}_3/\text{Al}$	<b>88/12</b>	<b>31</b>	<b>380–480</b>	<b>210–280</b>
	86/14	28	400–500	250–300
<b>Configuration I</b>	<b>77/18/5</b>	<b>35</b>	<b>400–500</b>	<b>270–320</b>
<b><math>\text{Bi}_2\text{O}_3/\text{Al}/1\text{Me-3H}</math></b>	<b>57/28/15</b>	<b>25</b>	<b>430–530</b>	<b>250–300</b>
Configuration II	83/10/7	36	410–510	150–200
$\text{Bi}_2\text{O}_3/\text{Al}/1\text{Me-3H}$	77/8/15	36	440–540	120–170
<b>Configuration III</b>	<b>79/11/10</b>	<b>39</b>	<b>450–550</b>	<b>170–220</b>
<b><math>\text{Bi}_2\text{O}_3/\text{Al}/1\text{Me-3H}</math></b>	<b>75/10/15</b>	<b>37</b>	<b>530–630</b>	<b>270–320</b>
Configuration IV	68/11/21	37	590–690	190–250
$\text{Bi}_2\text{O}_3/\text{Al}/1\text{Me-3H}$	73/12/15	37	570–670	380–430

Tube – the burning rate of the nanothermite composition tends to increase in the following sequence II<I<III<IV.

Thin layer– in the case of the ratio of the components of the mixture with max  $F$ , the tendency to increase the burn rate in the following row II<III<IV<I. For compositions with a content of 1Me-3H – 15%, there is a tendency to increase burn rate in sequence II<I<III<IV.

# PHOTOS OF TUBES AFTER MEASURING THE BURN RATE

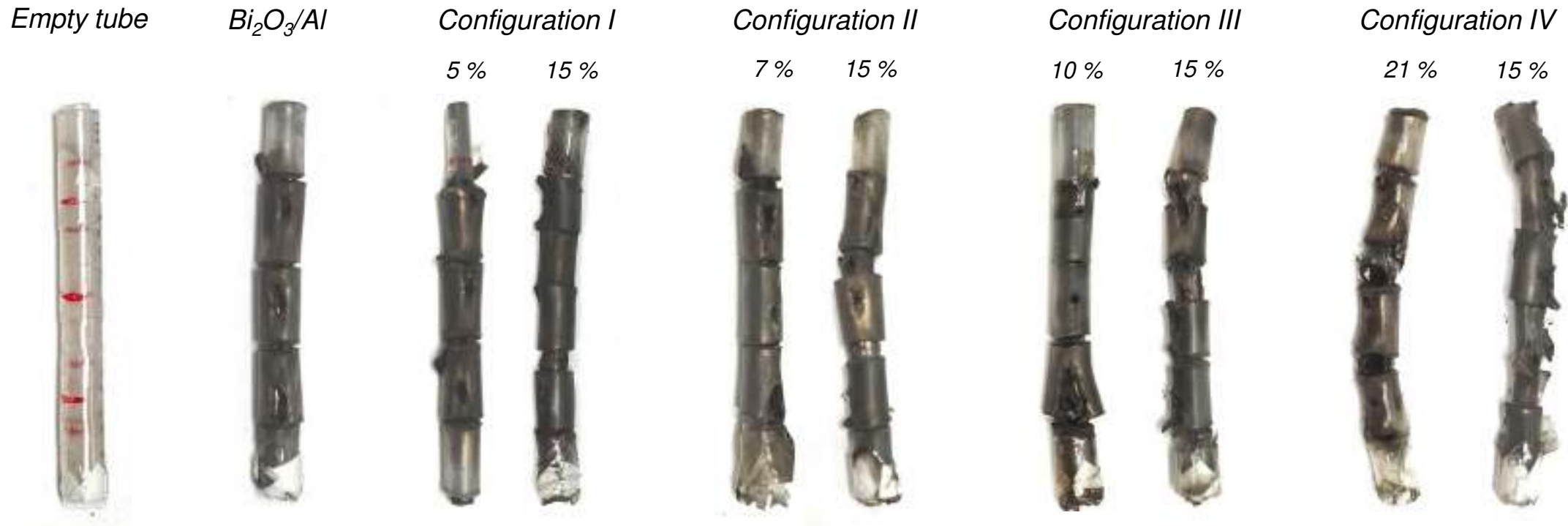


Fig. 5 – The appearance of the shells after burn rate measurement

As a result of the combustion composition  $\text{Bi}_2\text{O}_3/\text{Al}/1\text{Me}-3\text{H}$  the visually estimated degree of destruction of the shell changes in the following order  $\text{I} \leq \text{II} < \text{III} < \text{IV}$ , which allows us to talk about an increase in the «efficiency» of the nanothermic composition when changing the prescription configuration.



## CONCLUSION

1. A significant influence of the prescription configuration of the triple system  $\text{Bi}_2\text{O}_3/\text{Al}/1\text{Me-3H}$  on its combustion performance has been established.
2. Concentrations of 1Me-3H have been identified at which the maximum values of the explosion force are achieved with various compounding arrangements of the nanothermite composition  $\text{Bi}_2\text{O}_3/\text{Al}/1\text{Me-3H}$ : I – 122 % by 5 % 1Me-3H; III – 122 % by 10 % 1Me-3H; IV – 129 % by 21 % 1Me-3H. In the case of configuration II, the explosion force of the composition does not increase, only when the additive content is 7%, the  $F$  value remains at the level of the base mixture.
3. When changing the prescription configuration, there is a tendency to increase the burning rate of the nanothermic composition in the following sequence  $\text{II} < \text{I} < \text{III} < \text{IV}$ . The maximum values of the burning rate of the composition are characteristic of the configuration IV – 590-690 m/s in tube and 380-430 m/s in thin layer.

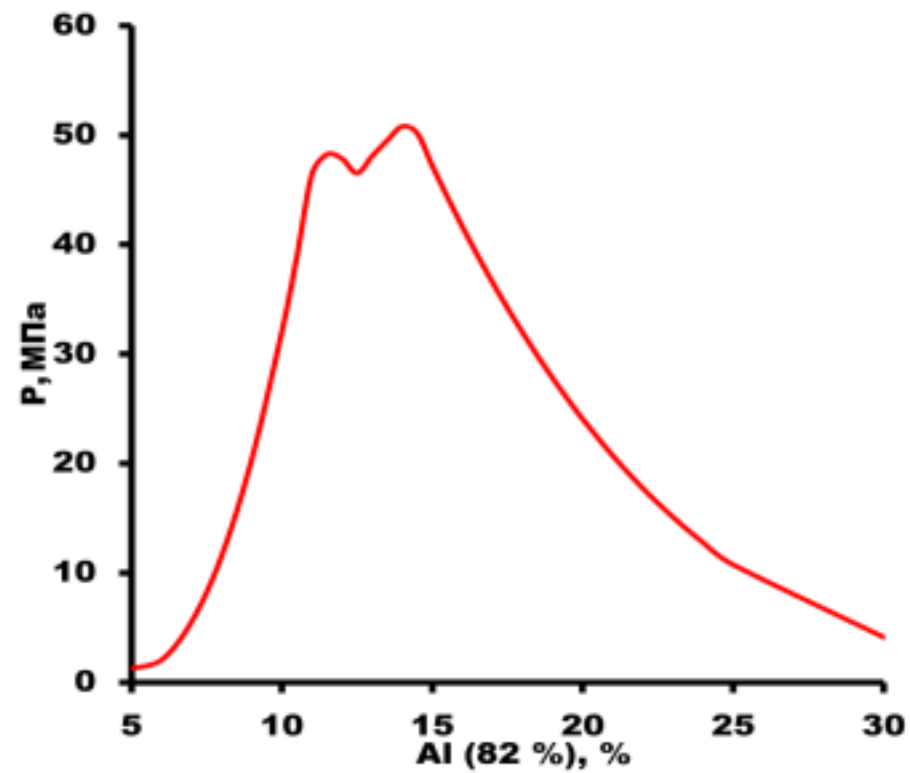


Fig. 6 – Dependence of the calculated pressure of the  $\text{Bi}_2\text{O}_3/\text{Al}$  nanothermite mixture on the Al content

When the aluminum content in the nanothermite mixture is 12-13%, a pressure drop is observed, in the case of configuration I (7 % 1Me-3H) and II (3, 5 % 1Me-3H), the ratio of the nanothermite pair in the composition  $\text{Bi}_2\text{O}_3/\text{Al}/1\text{Me-3H}$  falls into this area, which leads to a sharp decrease in the explosion force.