Electron spin resonance in situ study of Ni catalyst in catalytic transfer hydrogenation reaction conditions

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Ni-TiO₂ catalyst in the reaction

Ni-TiO₂ catalyst shows activity in the catalytic transfer hydrogenation reaction in supercritical i-PrOH media

It was shown, the catalytic activity increases with time as if the reduction of the catalyst happens [2]

in situ FMR reduction by H₂

Processes of the Ni nanoparticles formation can be studied by ferromagnetic resonance (FMR) in situ



Quartz reactor Temperature up to 1000 K Reduction/oxidation in situ Sensitive to nanoparticles size





in situ FMR, reactivation in SC i-PrOH

FMR in situ experiment on magnetic nanoparticles in the supercritical media





in situ, FMR reduction by SC i-PrOH Supercritical i-PrOH is capable of reducing NiO-TiO₂ sample without using gaseous H_2

Reduction temperature is 590 K, but the

In situ FMR reducing/passivation/reactivation of Ni-TiO₂ catalyst

Reduction of Ni-TiO₂ sample occurs at 600 K ((a)-(b)-(c))

Passivation in ~1% O_2 flow at 300 K leads to formation of NiO layer on the nanoparticles surface ((d)-(e))

Repeated reduction leads to the oxide layer removal ((e)-(f))

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