

New erbium (3+) hexafluoroacetylacetonates with spirocyclic photochromes: synthesis, structure, properties

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The functionalization of monomolecular magnets (MMM), viz metal complexes with various organic ligands, in order to give them additional (photo-, conductive, etc.) properties is a "hot topic" in the chemistry and physics of new materials. For example, the reversible photocyclization of complexes containing a photochromic component in the molecule upon UV or visible light irradiation can cause "switching on/switching off" the slow MMM relaxation [1].

Heavy lanthanide ions are increasingly being used for the design of MMMs due to their significant magnetic anisotropy and greater sensitivity to the surrounding crystal field [2]. It has been established that in a number of such molecular systems, the quantum tunneling mechanism of relaxation is controlled by photoisomerization [3, 4]. However, due to the high degree of coordination of 4f ions, systems with a weak modification of magnetic properties have been obtained so far [5–8].

In this work, new Er (3+) hexafluoroacetylacetonates with spirocyclic photochromes were synthesized. Their structure, magnetic properties, and photochemical activity were studied.

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- [1] A. Fetoh et al. *Scientific reports* **2016**, 6, 1, pp. 1-6.
- [2] M. Nakano, et.al., *Chem. Soc. Reviews* **2011**, 40, 6, pp. 3239-3248.
- [3] D. Pinkowicz, et al. , *Chemistry –A Eur. J.* **2014**, 20, 39, pp. 12502-12513.
- [4] G. Cosquer, et al., *Dalt. Trans.* **2015**, 44, 13, pp. 5996-6002.
- [5] G. Cosquer, et al., *Inorganics* **2018**, 6(1), p. 9.
- [6] D. Pinkowicz, et al., *Chem. Eur. J* **2014**, 20, pp. 12502–12513.
- [7] G. Cosquer, et al., *Dalt. Trans.* **2015**, 44, pp. 5996–6002.
- [8] A. Fetoh, et al., *Inorg. Chemistry* **2019**, 58 (4), pp. 2307-2314.