

UV-photoexcitation of oxygen-isoprene collision complexes as a source of singlet oxygen

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The interaction of oxygen molecules with molecular environment provides the strong enhancement of UV-absorption by oxygen and other molecules which takes place when molecules collide in the gas phase. Thus, isoprene C₅H₈ (the most widespread biogenic organic compound in the Earth atmosphere after methane), being transparent at wavelengths $\lambda > 240$ nm, absorbs in the presence of molecular oxygen due to collision-induced absorption in C₅H₈-O₂ collision complexes [1].

Oxygen collision-induced absorption also dramatically changes oxygen photochemistry causing new photochemical processes such as formation of singlet oxygen species ¹O₂ which possess high chemical reactivity and play important role in nature. Recently we have observed ¹O₂ formation as result of the isoprene-oxygen collision complexes photoexcitation in the gas phase within the spectral region 253.5–355 nm at the oxygen elevated pressure [2]. Singlet oxygen has been detected with its NIR luminescence centered near 1.27 μ m. The ¹O₂ photogeneration is found to be a one-photon process. We have measured the ¹O₂ quantum yield within the UV-C region (253–278 nm) and supposed it is governed mainly by O₂ molecules photoexcitation to the Herzberg III (³ Δ_u) state via C₅H₈-O₂ collision complexes enhanced absorption. So excited triplet O₂ gives rise to ¹O₂ because of triplet-triplet annihilation in the collisions with unexcited O₂ molecules. In the UV-B (308 nm) region the ¹O₂ appearance is attributed to the excitation of a double spin-flip (DSF) transition in C₅H₈-O₂ complex. In the UV-A region (355 nm) besides DSF isoprene sensitizes ¹O₂ formation as result of O₂-assisted excitation to the triplet state. We suppose that DSF may result in ¹O₂ formation in widespread wavelength region depending on oxygen collisional partner X in any oxygen-containing media. Relying on the obtained data we made estimations of the role of this new process in the Earth atmosphere.

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[2] A.P. Pyryaeva, K.S. Ershov, S.A. Kochubei, A.V. Baklanov, *J. Phys. Chem. A* **2020**, 124, pp. 8469–8477.